

Polymer-Enhanced Cs₃Cu₂I₅ Perovskite Scintillator Films with Improved Stability and X-Ray Imaging Quality

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Scintillators convert high-energy radiation such as X-rays into visible light and are essential for applications in medical imaging, industrial non-destructive testing, and security screening. However, conventional scintillators—including Bi₄Ge₃O₁₂, CdWO₄, BaF₂, CsI:Tl, and NaI:Tl—are limited by high-temperature single-crystal growth, toxicity or rare-earth element incorporation, and insufficient chemical stability.[1]

Halide perovskites have recently emerged as attractive candidates for next-generation scintillators due to their strong absorption of high-energy photons, fast emission dynamics, and structural tunability. **Copper-based halide perovskites** adopt zero-dimensional (0D) crystal structures that promote strong lattice distortion and the formation of **self-trapped excitons (STEs)**. This results in broadband emission with a **large Stokes shift**, effectively suppressing self-absorption—one of the primary efficiency-limiting factors in conventional and lead-based perovskite scintillators.

Owing to their STE-mediated photophysics and intrinsically non-toxic composition, copper-based halide perovskites exhibit highly efficient scintillation with excellent quantum yields. These results highlight copper-based halide perovskites as a promising, environmentally benign scintillator platform for radiation detection in medical, industrial, and security applications.

1. Li, N., Xu, Z., Xiao, Y., Liu, Y., Yang, Z., & Liu, S., “Flexible, High Scintillation Yield Cu₃Cu₂I₅ Film Made of Ball-Milled Powder for High Spatial Resolution X-Ray Imaging,” *Adv. Optical Mater.*, **10**, 2102232 (2022).

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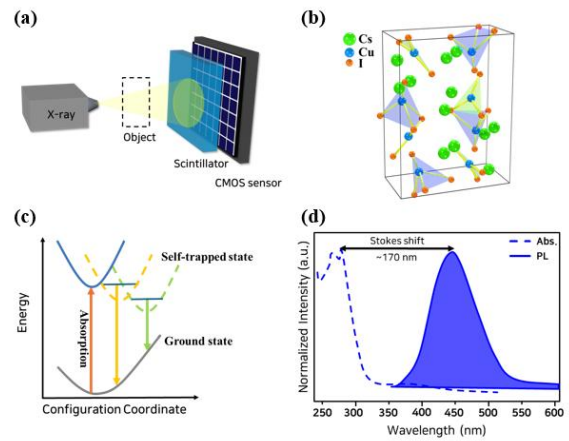


Figure 1. (a) Schematic illustration of a scintillator-based X-ray imaging system, (b) crystal structure and optical properties of Cs₃Cu₂I₅, (c) formation of self-trapped excitons (STEs) induced by lattice distortion in copper-based halide perovskites, and (d) the resulting large Stokes shift that suppresses self-absorption.