



# Tb<sup>3+</sup> Doped Silicoborate Glass Scintillator for High Resolution Synchrotron X-Rays Imaging Application

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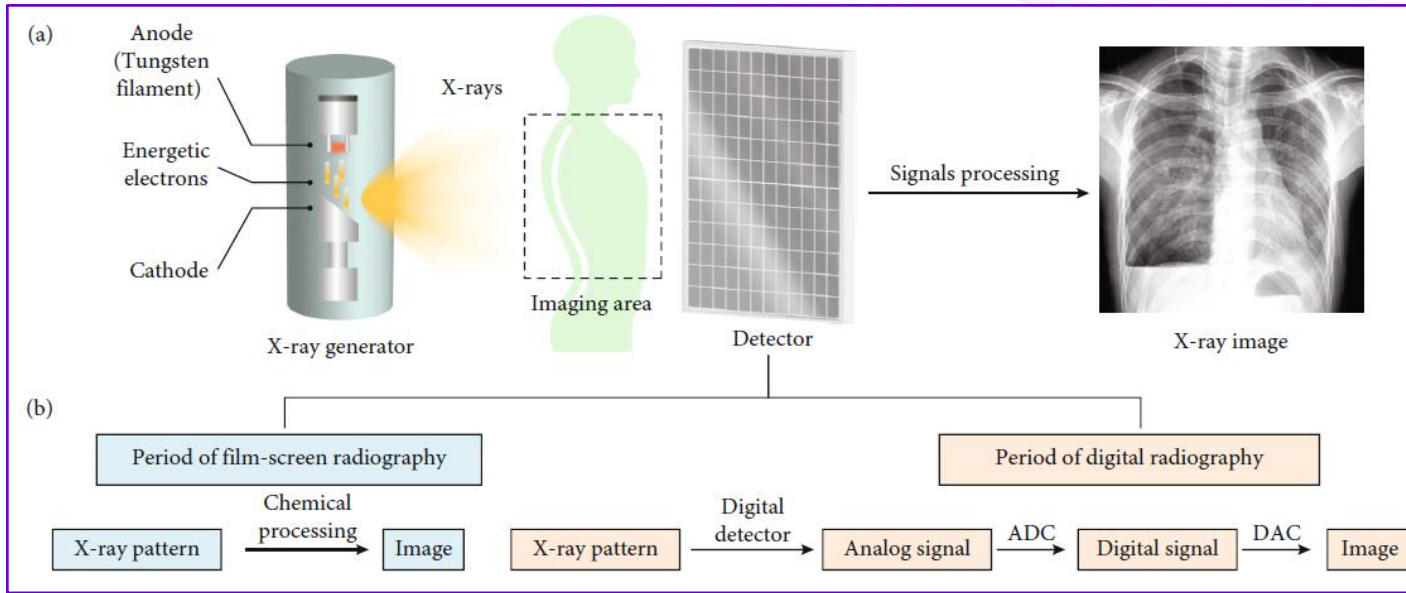
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# Research Background & Motivation



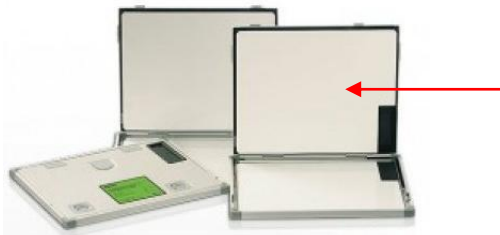
**Figure 1:** (a) Schematic illustration of an X-ray imaging system. (b) The development of X-ray radiography with the evolution of X-ray detectors.

**Radiography** is a medical imaging technique that uses X-rays to create images of the internal structures of the body.

- It is commonly used for diagnostic purposes to visualize bones, organs, and tissues.
- The X-ray machine emits a controlled amount of radiation, which passes through the body and is captured on a film or digital detector on the other side.

Research (2021), Article ID 9892152, <https://doi.org/10.34133/2021/9892152>

## • Intensifying screen



### Phosphor

- $\text{CaWO}_3$

### Rare earth elements

- $\text{La}_2\text{O}_2\text{S}$
- $\text{LaBrO}$
- $\text{Gd}_2\text{O}_2\text{S}$

<https://pjaxray.com/green-400-speed-screens-and-cassettes-1417.html>

## • Computed radiography (CR)



### Photostimulable phosphor (PSP) plate

- $\text{BaFBr:Eu}^{2+}$
- $\text{CsBr:Eu}^{2+}$

<https://www.indiamart.com/proddetail/fuji-x-ray-cr-cassette-23371763791.html>

## • Digital radiography (DR)



### Material of scintillators

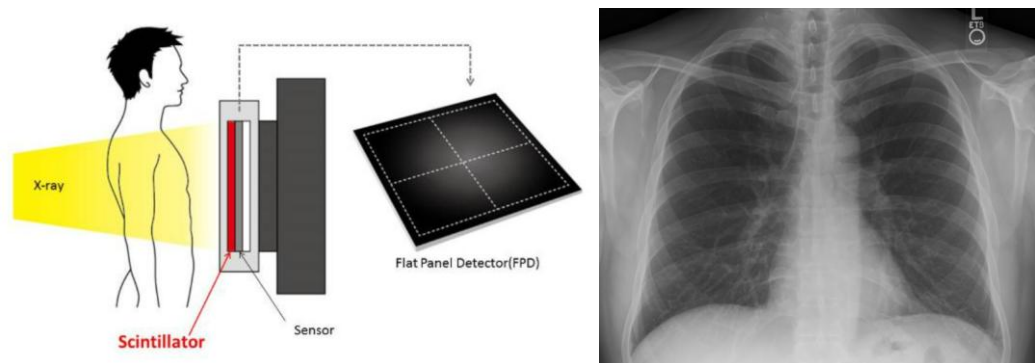
- $\text{CsI:TI}$
- $\text{GOS: Tb}$  (Terbium doped Gadolinium oxysulfide)

<https://www.tramed.com/products/8577/Flat-Panel-Detector.html>

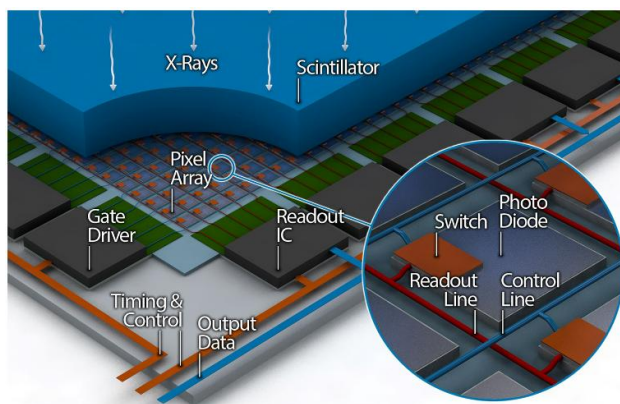
# Research Background & Motivation

## Detector in General Radiography

### • General Radiography



### • Flat Panel Detector



**Scintillators** – is a materials that spontaneously emit light at wavelengths within or near the visible region of the electromagnetic spectrum upon exposure to ionizing radiation

<https://www.cappa.ie/photonic-applications/>  
<https://ams.com/how-flat-panel-detectors-work>

## Types of scintillator

Types of scintillator	Advantages	Disadvantages
• Gas scintillator $^3\text{He}$	Excellent $\gamma$ /neutron discrimination	Expensive
• Single-crystal scintillators	High light yield	Difficult to grow
• Plastic scintillators	Easy to form	Low light output
	- Can be variety of size and shapes	
• <b>Glass scintillators</b>	- <b>Doped with composition or lanthanide ions into the glass matrix; improve radiation absorption &amp; tune emission wavelengths</b>	- Quite fragile and easily damaged - Light output weak; added activator to solve

J.D. Musgraves, et al. Handbook of glass 2019

# Research Background & Motivation

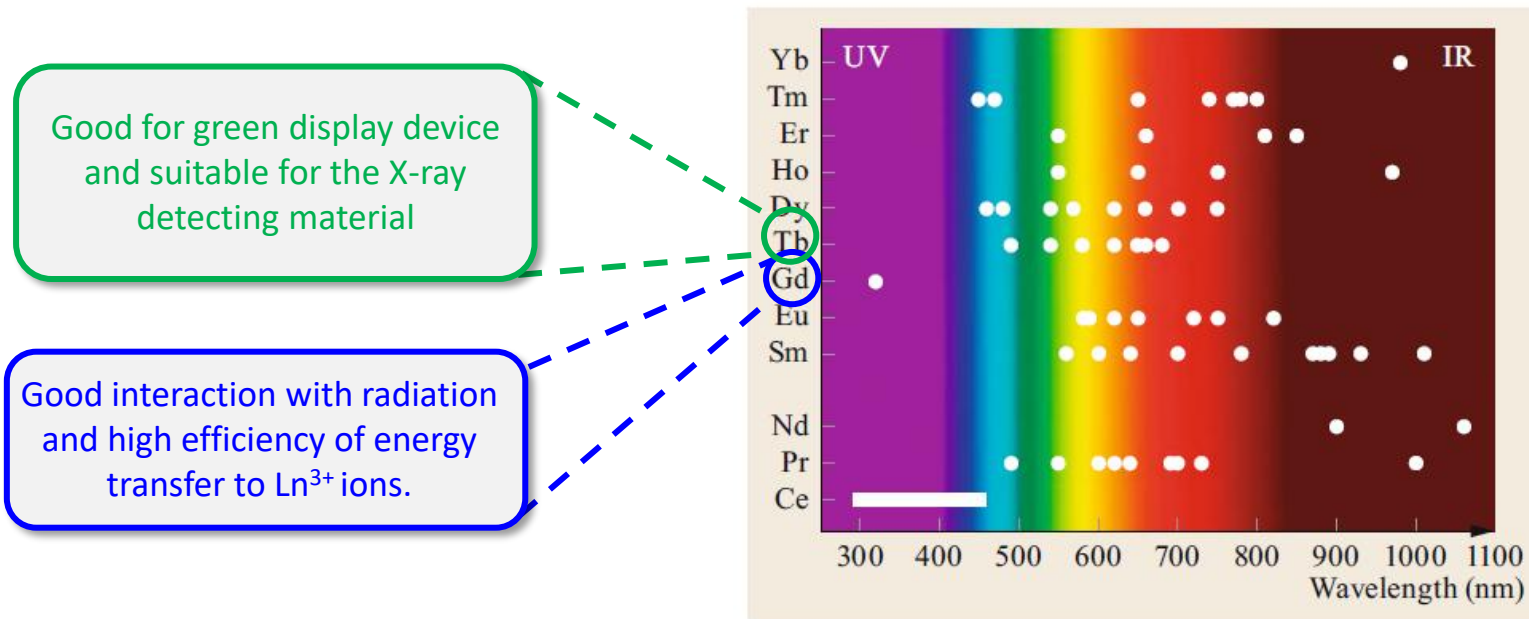
## ■ Lanthanides in the periodic table with atomic number and weight

58	59	60	61	62	63	64	65	66	67	68	69	70	71
Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu
140.115	140.9076	144.24	(145)	150.36	151.965	157.25	158.9253	162.50	164.9303	167.26	168.9342	173.04	174.967

## ❖ Dopants, which serve as activators

create the energy gap in luminescence center resulting to..

- ✓ Increase efficiency
- ✓ Control emission characteristics
- ✓ Enhance luminescence properties



Wavelengths of the main emissive transitions of the trivalent lanthanide ( $\text{Ln}^{3+}$ ) coordination compounds in the near-UV to NIR range.

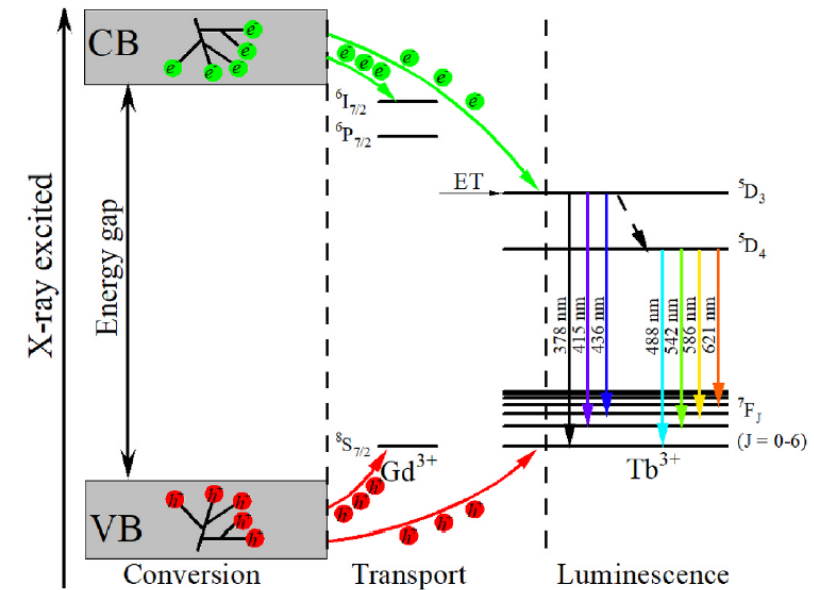


Fig. 6. Schematic diagram scintillating mechanism for ABSG:xTb glasses, where VB denotes valance band and CB denotes conduction band.

## References

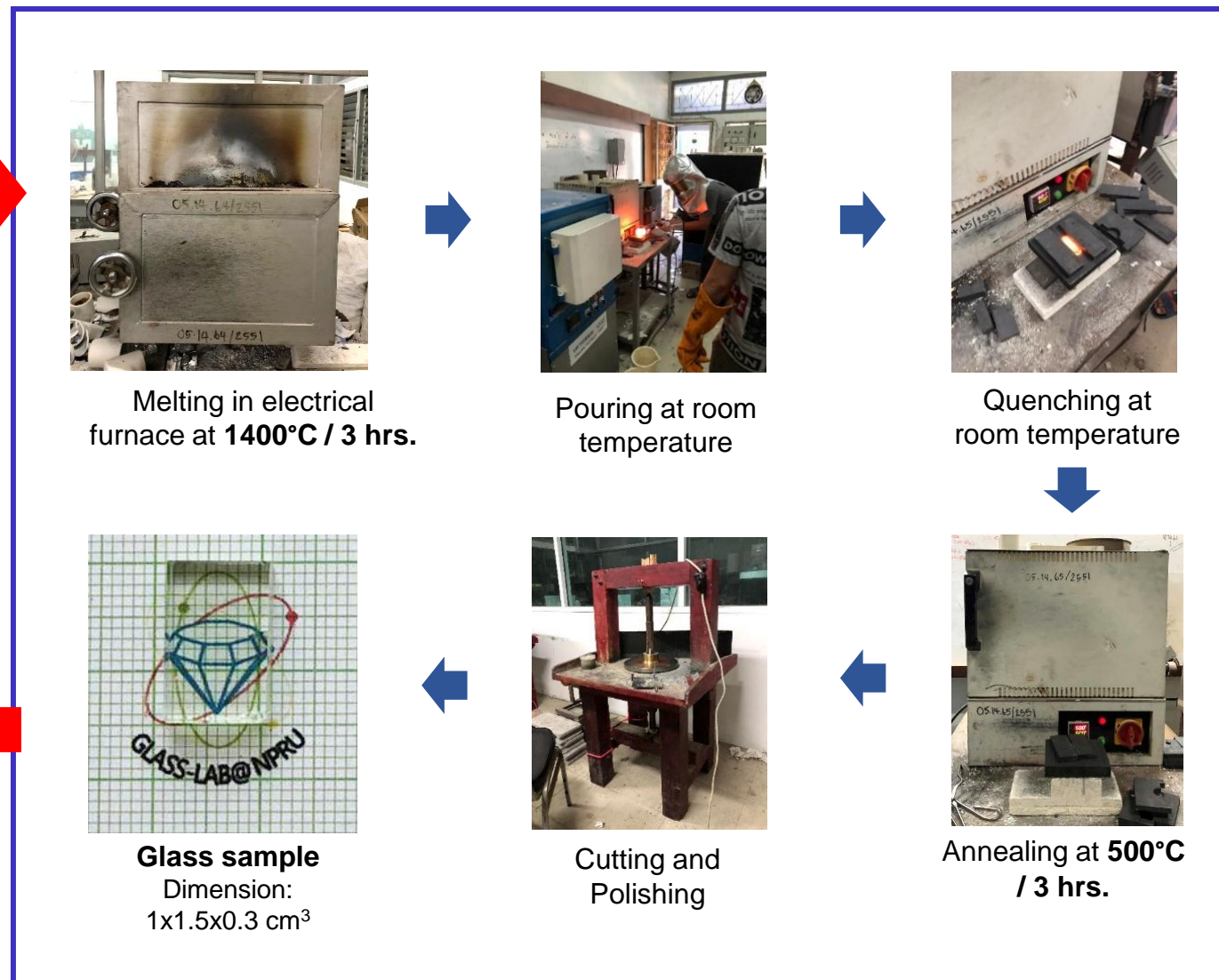
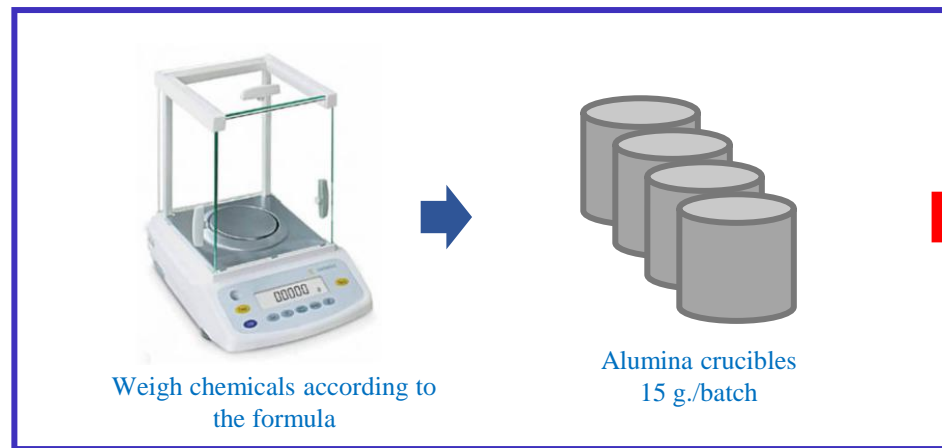
- J. David Musgraves, et al., Springer Handbook of Glass, 2019, <https://doi.org/10.1007/978-3-319-93728-1>  
 Palvi Gupta, et al., 2015, Mater. Res. Express, <https://doi.org/10.1088/2053-1591/2/7/076202>  
 WenJun Huang, et al., Ceramics International 48 (2022) 17178–17184, [doi.org/10.1016/j.ceramint.2022.02.274](https://doi.org/10.1016/j.ceramint.2022.02.274)

## Objectives

- To develop Tb<sup>3+</sup> doped silicoborate glass to be used as a glass scintillator.
- To investigate the physical, optical, luminescence properties of Tb<sup>3+</sup> doped silicoborate glass.
- To study the possibility of glass to be used as a scintillator, carried out based on the “Synchrotron radiation X-ray tomographic microscopy (SRXTM)” technique.

# Methodology

□ **Glass formula:**  $xTb_2O_3 : 40Na_2O : 7.5Gd_2O_3 : 5SiO_2 : (47.5-x)B_2O_3$  ( $x = 0, 1, 2$  and  $3$  mol%)



□ **Investigation its property of...**

- **Physical and optical properties**

- Density, Molar volume
  - Absorption spectra

- **Luminescence properties**

- Photoluminescence (PL) → Excitation & Emission spectra, Decay time
  - X-ray induced luminescence
  - PLQY

- **X-ray imaging**

# Research Results

## Physical and optical property

### ❖ Density and molar volume

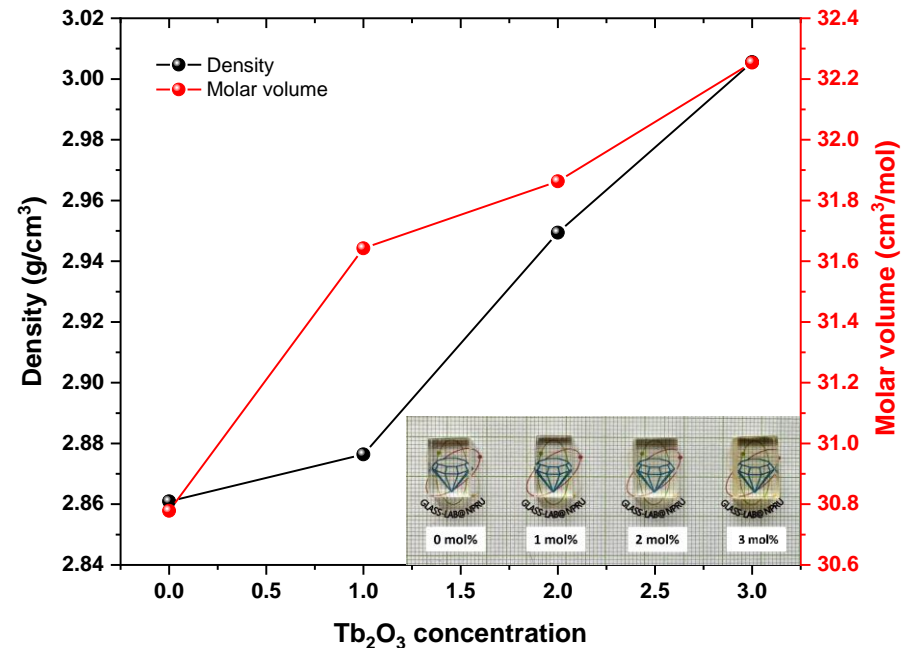


Fig.1. Density and Molar volume of xTb:7.5Gd glasses

### ❖ Absorption spectra measured by UV-Vis-NIR spectrometer

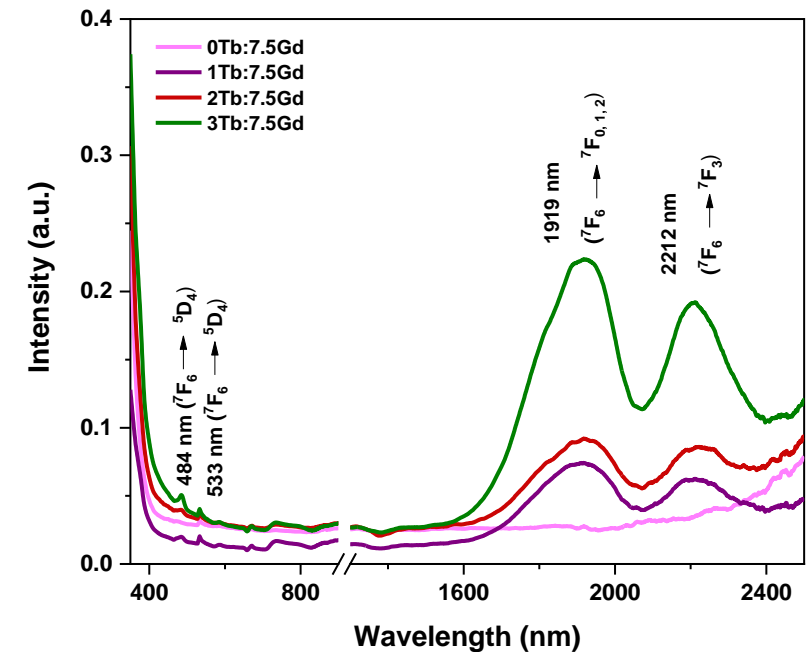


Fig.2. Vis-NIR absorption spectra of xTb:7.5Gd glasses

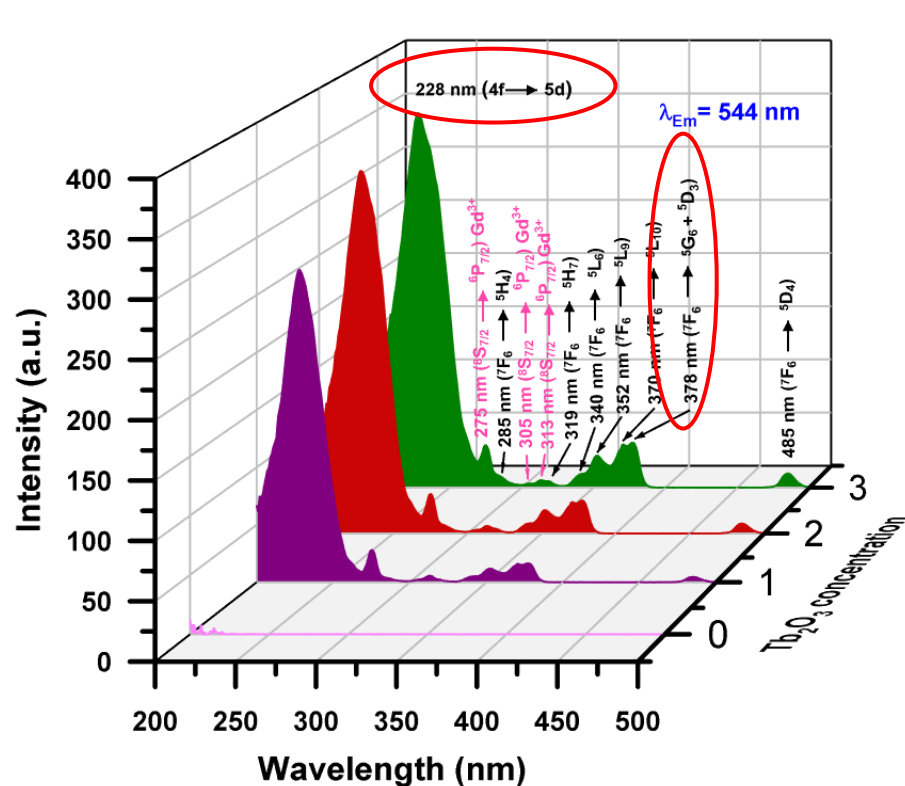
- **Density and molar volume** increases with Tb<sub>2</sub>O<sub>3</sub> concentrations increasing; B<sub>2</sub>O<sub>3</sub> components were replaced by Tb<sub>2</sub>O<sub>3</sub>.
- Tb<sup>3+</sup> can absorb the light at Vis and NIR region. **Absorption spectra** of these wavelengths **increased** with increasing concentration of Tb<sub>2</sub>O<sub>3</sub>.



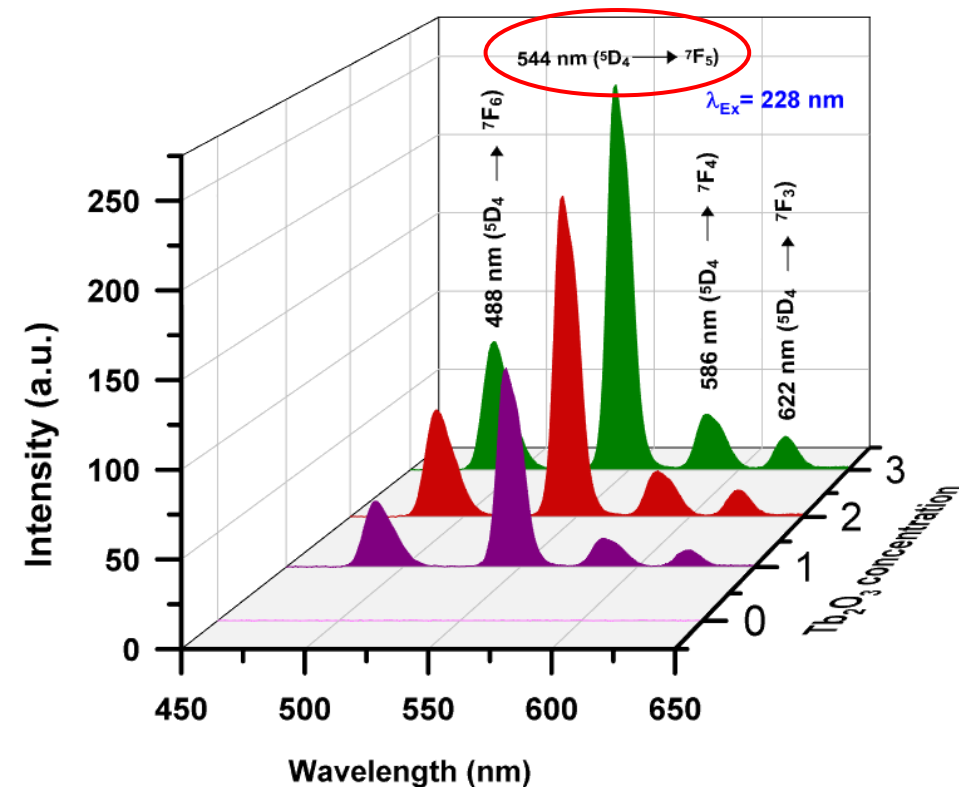
# Research Results

## Luminescence property

❖ Photoluminescence spectra measured by Agilent Cary Eclipse fluorescence spectrometer



**Fig.3.** Excitation spectra of xTb:7.5Gd glasses under  $\lambda_{Em} = 544$  nm



**Fig.4.** Emission spectra of xTb:7.5Gd glasses under  $\lambda_{Ex} = 228$  nm

- The emission spectra under 228 nm excitation showed the **strong green emission peaks at 544 nm** corresponding to  $^5D_4 \rightarrow ^7F_5$  transition in 3.0 mol% of  $Tb^{3+}$ -doped glass.

# Research Results

## Luminescence property

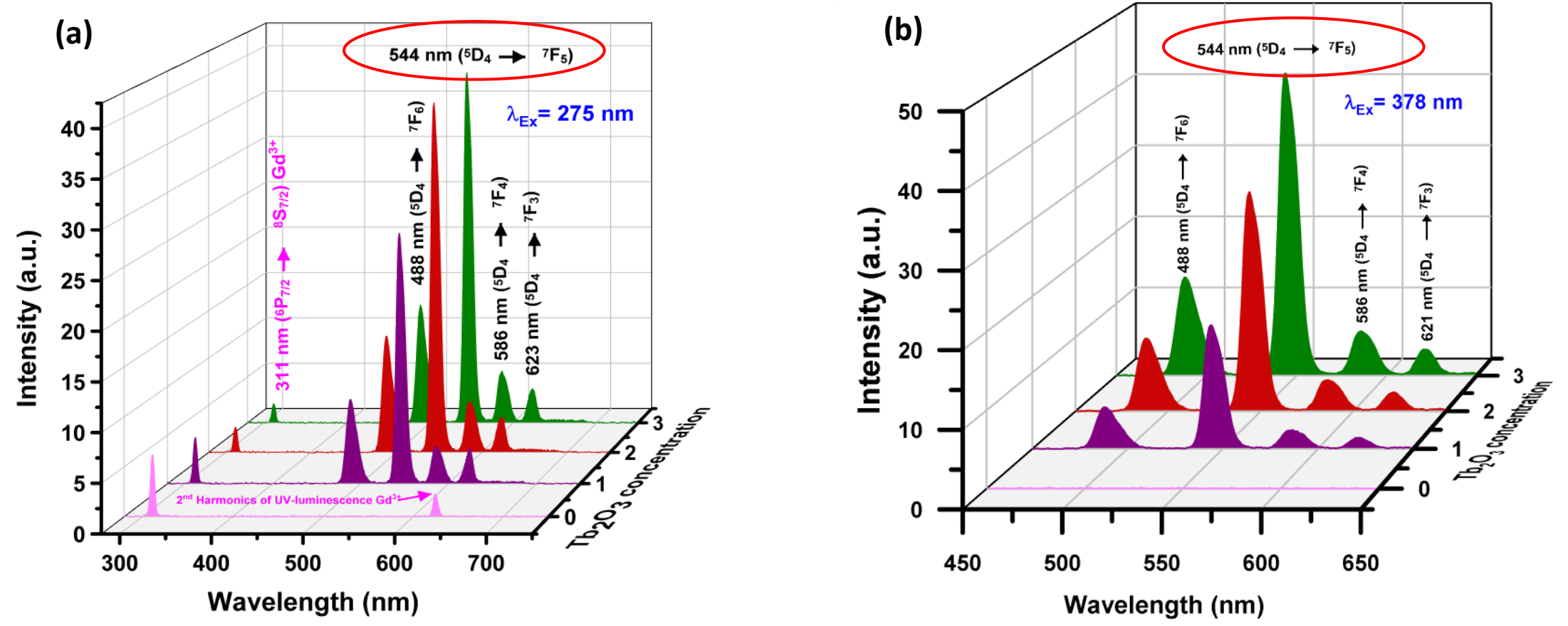


Fig.5. Emission spectra of xTb:7.5Gd glasses under  $\lambda_{Ex} = 275$  nm (a) and 378 nm (b)

- The additional peak centering at 311 nm describes the **Gd<sup>3+</sup> transition** emission under  ${}^6P_{7/2} \rightarrow {}^8S_{7/2}$  while the additional peak at 623 nm is the **half-harmonic** of 275 nm.

# Research Results

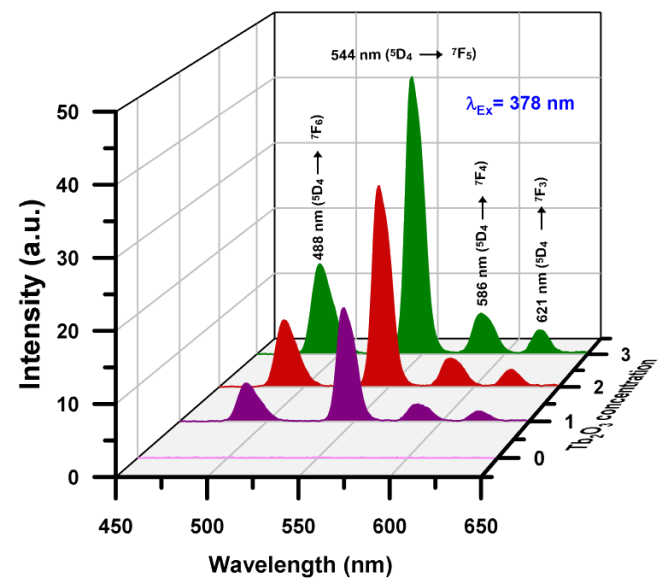
## Experimental photoluminescence quantum yield (PLQY)

- ❖ Photoluminescence quantum yield was utilized by spectrofluorometer with an excitation source of a xenon lamp and K-integrating sphere

**Table 1**

The experimental photoluminescence quantum yield (PLQY) of xEu:7.5Gd glasses under  $\lambda_{Ex} = 378$  nm at the wavelength of 400 – 650 nm.

Glass	Experimental PLQY (%)	Abs Error	Re Error
1Tb:7.5Gd	22.65	-	-
2Tb:7.5Gd	26.79	0.108	0.004
3Tb:7.5Gd	30.85	0.081	0.003

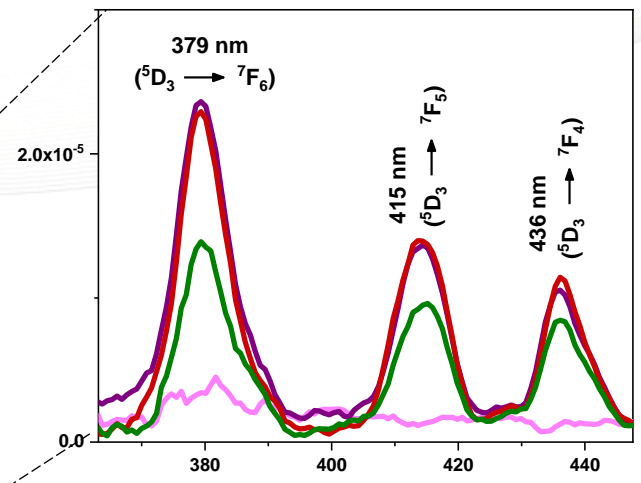
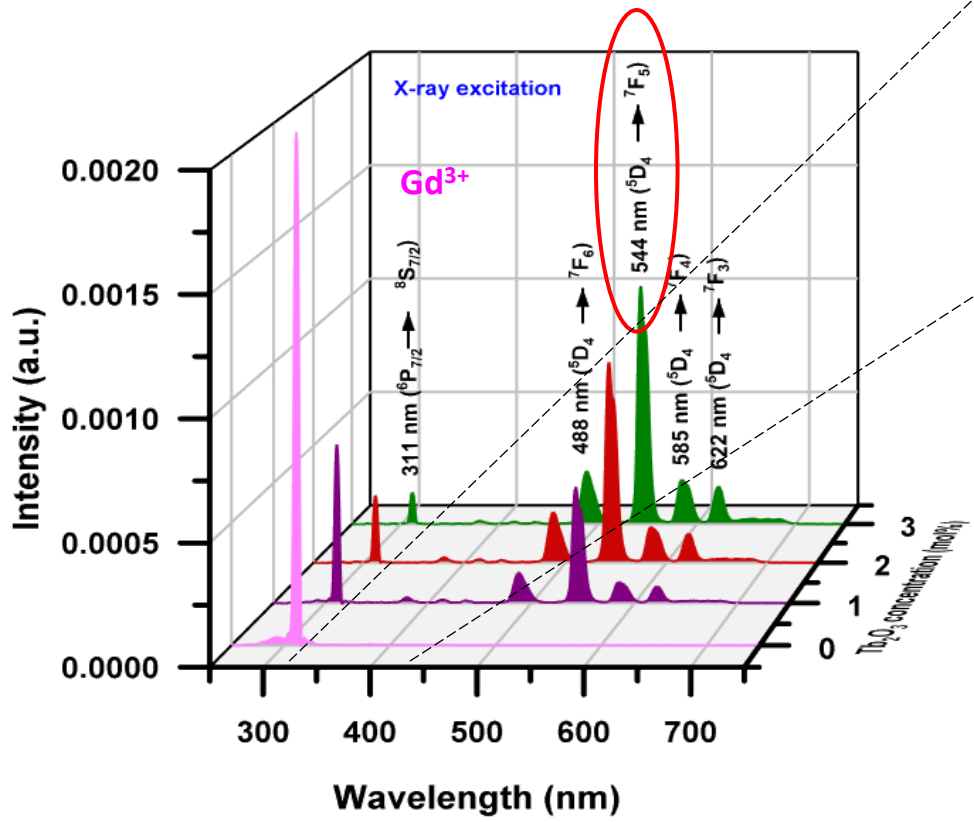


- This represents high-quality luminescence materials and indicates efficient light emission for a given amount of absorbed energy.

# Research Results

## X-ray excitation spectra

❖ X-ray induced luminescence spectra irradiated by X-ray, setting as 50 kV and 30 mA at NPRU, Thailand



➤ The emission intensity at  $^5D_4$  increases with adding  $Tb_2O_3$  concentrations while  $^5D_3$  decreased

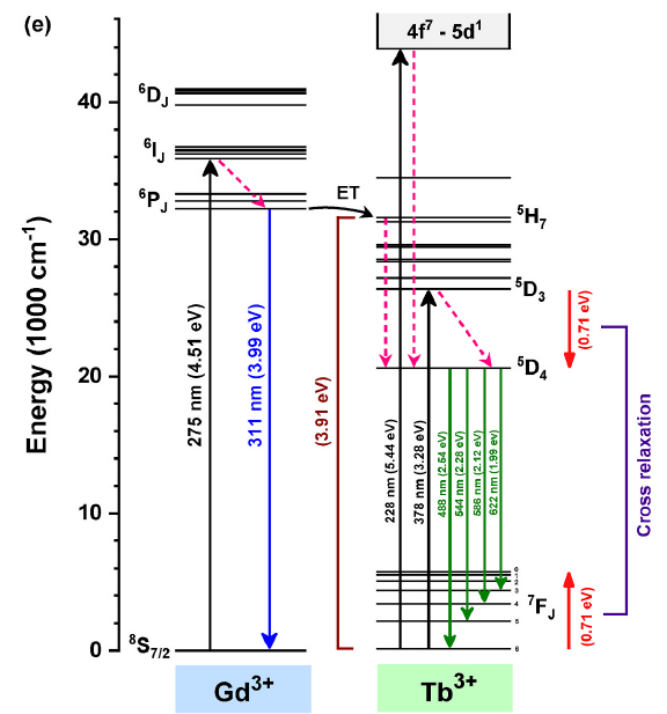


Fig.6. X-ray excitation spectra of xTb:7.5Gd glasses

# Research Results

## X-ray excitation spectra and CIE chromaticity diagram

❖ Comparing the total integrated area of 3Tb:7.5Gd glass and BGO

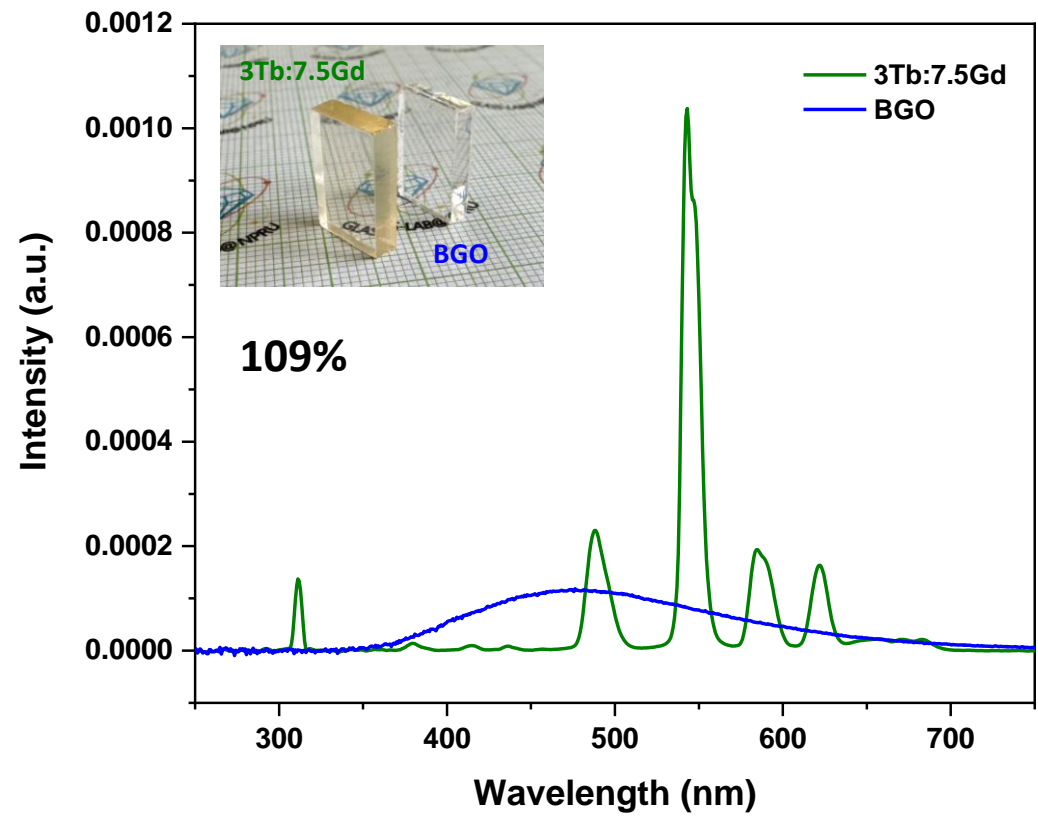


Fig.7. X-ray excitation spectra of 3Tb:7.5Gd compared with BGO

❖ The color of emitted light investigated by the International Commission on Illumination (CIE 1931), using emission data

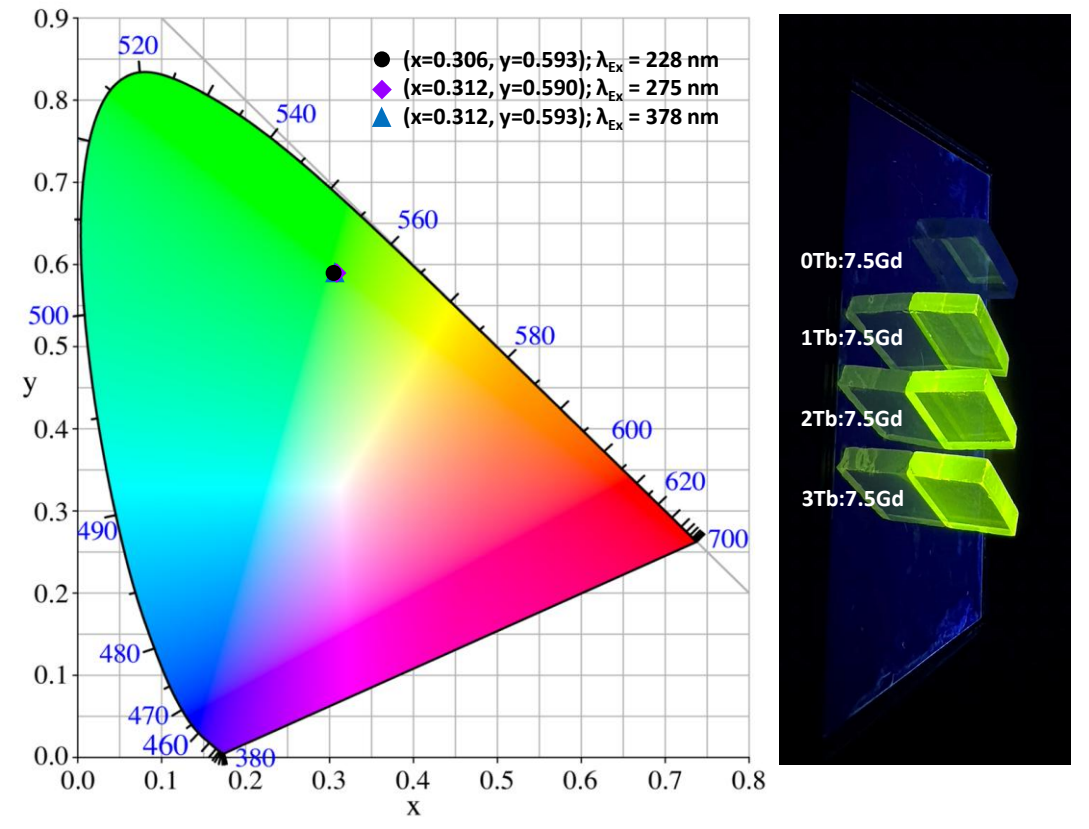
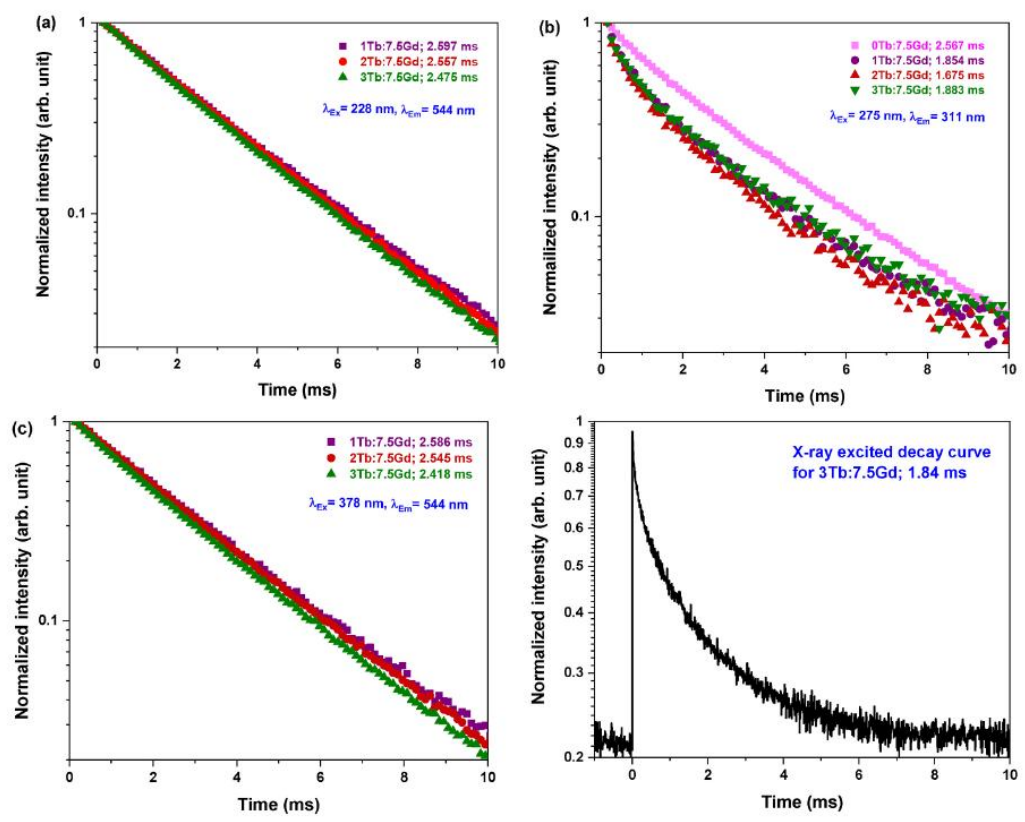


Fig.8. CIE chromaticity of xTb:7.5Gd glasses

# Research results

## Decay curves and Energy transfer efficiency

### Decay curves



**Fig.9.** Decay time of the xTb:7.5Gd glasses, excitation at (a) 228 nm (b) 275 nm (c) 378 nm and (d) X-rays excitation.

### Energy transfer efficiency ( $\eta_{ET}$ )

$$\eta_{ET} = 1 - \frac{\tau_{Gd-Tb}}{\tau_{Gd}}$$

$\tau_{Gd-Tb}$  = the decay time of the donor ( $Gd^{3+}$ ) in the presence of the acceptor ( $Tb^{3+}$ )  
 $\tau_{Gd}$  = the decay time of the donor in the absence of the acceptor

**Table 2** The emission decay time ( $\tau$ ) under  $\lambda_{Ex}=275$  nm,  $\lambda_{Em}=311$  nm and energy transfer efficiency ( $\eta_{ET}$ ) for xTb:7.5Gd glasses

Glass sample	$\tau$ (ms)	$\eta_{ET}$ (%)
( $\tau_{Gd}$ )		
0Tb:7.5Gd	2.567	-
( $\tau_{Gd-Tb}$ )		
1Tb:7.5Gd	1.854	28
2Tb:7.5Gd	1.675	35
3Tb:7.5Gd	1.883	27

# Research Results

**X-ray imaging** ; Experiment at BL1.2W: X-ray Imaging & X-ray Tomographic Microscopy (XTM), Synchrotron Light Research Center, Nakhon Ratchasima, Thailand

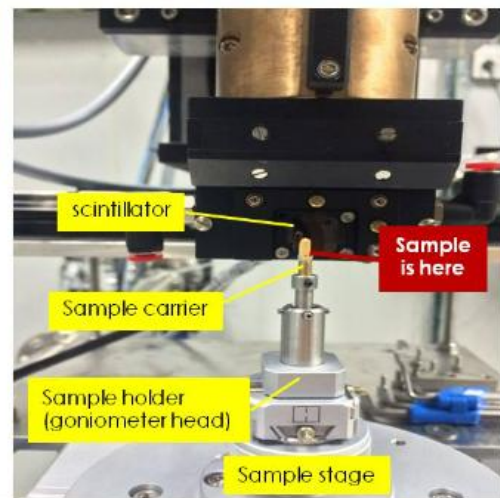
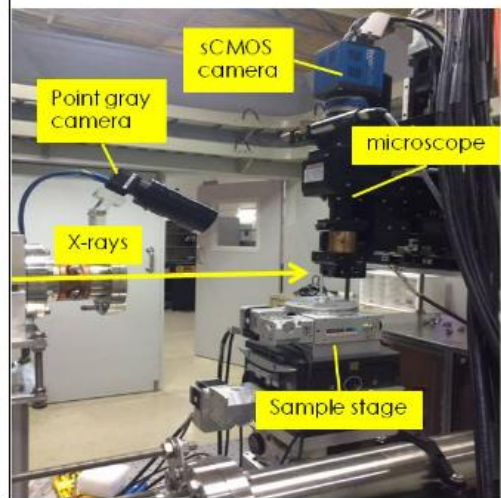


Figure 3 Experimental hutch of BL1.2W (commissioning mode).

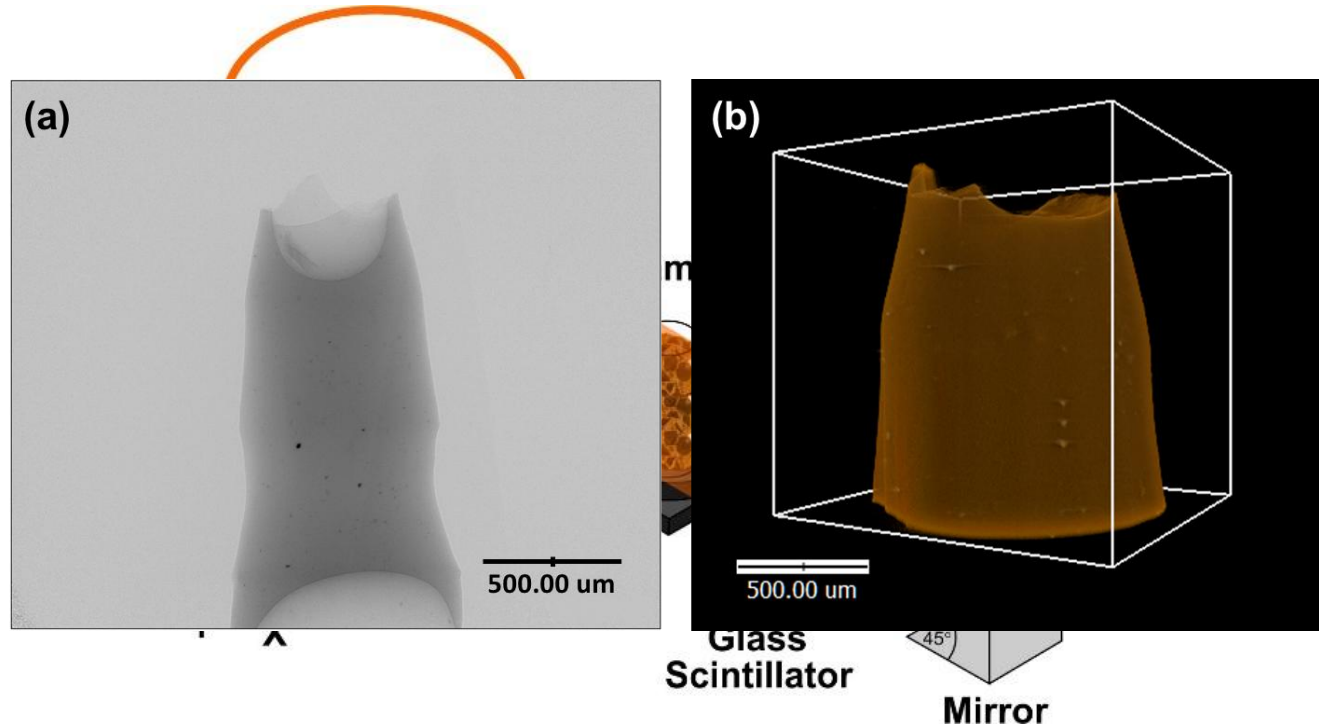
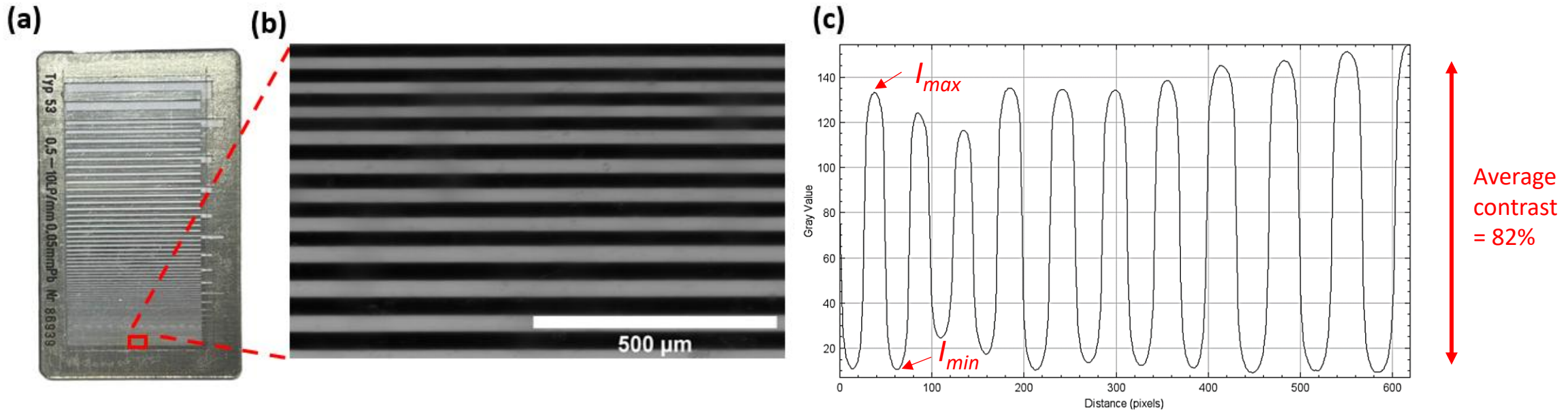


Fig.11. The 2D X-ray image (a) and the 3D reconstruction from 2D X-ray images of slime (b) by 3TbEu:7.5Gd glass as a scintillator.

# Research Results

## X-ray imaging

- ❖ **Spatial resolution** can be expressed using spatial frequency, which describes the number of line pairs in a certain length (lp/mm).



**Fig.12.** Line pair pattern (a) and the 2D X-ray image of line pair pattern at 10 lp/mm (b) and plot profile (c) of 3Tb:7.5Gd glass as a scintillator.

$$\% \text{ Contrast} = \left[ \frac{I_{max} - I_{min}}{I_{max} + I_{min}} \right]$$

where  $I_{max}$  is the maximum intensity of gray value and  $I_{min}$  is the minimum intensity gray value



## xTb:7.5Gd glass

### Physical and Optical property

- Developed glass is heavy up to 3.01 g/cm<sup>3</sup>.
- Glass performs the **enhanced non-bridging oxygen** with increment of Tb<sub>2</sub>O<sub>3</sub> content.
- Tb<sup>3+</sup> doped glass absorbs photons in **Vis and NIR region**.

### Luminescence property

- PL and RL under all excitation makes the **strongest emission of Tb<sup>3+</sup> at 544 nm** (<sup>5</sup>D<sub>4</sub> → <sup>7</sup>F<sub>5</sub>) in 3.0 mol% doped glass.
- **PLQY** of 3Tb:7.5Gd was 30.85%
- Photoluminescence **decay curve** of Eu<sup>3+</sup> is in **millisecond order**.

### X-ray imaging

- **Special resolution** of X-ray imaging is **10 lp/mm**
- **Possibility to be a glass scintillator**, confirmed by X-ray imaging

# Acknowledgment



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Researcher at SLRI : Dr. Phakkhananan Pakawanit



**Thankyou**  
*for your kind attention*