# Catalysts

# for Environment, Energy and Health

# Prof. IN, SU-IL

Dept. of Energy Science & Engineering

DGIST



KAIST

DGIST

DGIST is a research-oriented university with a full support from Korean government.

DGIST started with a research institution in 2004 and grew into a researchoriented university by opening graduate program in 2011, and the undergraduate program in March, 2014. DGIST focuses its research and education in six areas, such as Emerging Materials Science, Information & Communication Engineering, Robotics Engineering, Energy Science and Engineering, Brain Science, and New Biology.

# **Microbial Fuel Cell for H<sub>2</sub>**



Photocoupled Bioanode: A New Approach for Improved Microbial Fuel Cell Performance

Energy Technology, 6(2), 2017



# **CO<sub>2</sub> Utilization**



High-Rate Solar-light Photoconversion of CO<sub>2</sub> to Fuel : Controllable Transformation from C<sub>1</sub> to C<sub>2</sub> Products

CO<sub>2</sub>, Water, and Sunlight to Hydrocarbon Fuels: A Sustained Sunlight to Fuel (Joule-to-Joule) Photoconversion Efficiency of 1%

Energy & Environ. Sci. (2019) 12, 2685 - 2696



## **Nuclear Power Battery**



C-14 Powered Dye-Sensitized Betavoltaic Cells

Chemical Communications (2020) 56, 7080-7083



**Nano-Bio Hybrid Technology** 



Multiplex Protein Imaging with Secondary Ion Mass Spectrometry Using Metal Oxide Nanoparticle Conjugated Antibodies



Enhanced Therapeutic Treatment of Colorectal Cancer Using Surface-Modified Nanoporous Acupuncture Needles

Scientific Reports 7 12900 (2017)

# **Topic 1 : Solar Light Driven CO<sub>2</sub> Reduction into Fuels**



- Energy Environ. Sci. (2018) 11, 3183 319
- Energy Environ. Sci. (2019) 12, 2685 2696

(2020) in press

- Applied Catal. B (2020) 279, 119344
- Chem. Eng. J.

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# **CO<sub>2</sub> Conversion into Fuels ?!**







# Nano material synthesis for green energy application

- CO<sub>2</sub> conversion into organic compounds, K. Honda et at., *Nature* 1979, 277, 637-638.
- Visible light photocatalysis in N-doped TiO<sub>2</sub>, R. Asahi et al., *Science*, 2001, 293, 269-271.
- Formation of hollow nanocrystals, P. Alivisatos et al., Science 2004, 304, 711-714.
- Ideal combination for CO<sub>2</sub> conversion, C. Grimes et al., ACS NANO, 2010, 4, 1259-1278.





 $TiO_2$  as a reference and standard material: non-toxicity, high photoactivity, mechanical stability, low cost, favorable overlap with the UV portion of the solar spectrum

# **Reactor Design for Photocatalytic Gas Phase Reaction**



2004~2008

Chem. Comm. 40 (2006) 4236 - 4238 JACS 129 (45) (2007) 13790-13791 Energy & Environ. Sci. 2(12) (2009)1277-1279.



2009~2010

JPC C, 2010, 114 (25), 11162-11168 Chem. Comm. 2011, 47 (9), 2613 - 2615 J Photochem. Photobio. A 2011, 222(1), 258-262 J Catalysis, 2012, 289, 62-72





Solar-thermal reactor

Angew. Chem. Inter. Ed. 51 (2012) 3915- 3918 Rapid Comm. Photosci. 2013, 2 (2), 64-66 ACS Omega, (2016) 1, 868-875 Carbon 98 (2016) 537-544. Journal of CO2 Utilization 20 (2017) 301-311

2012 ~

# **Gas Phase Flow Reactor System**





Angew. Chem. Int. Ed. 51 (2012) 3915- 3918



RSC Adv., 2016, 6, 38964–38971



ACS Omega 1 (2016) 868-875





J. of CO<sub>2</sub> Utilization 20 (2017) 91–96

### **Distinct Features of Reduced Titania**

- > Surface defects, oxygen vacancies ( $V_0$ ) and Ti<sup>3+</sup> ions:
  - Responsible for new shallow or deep mid gap energy states
  - Also acts as electron/hole traps rather than a recombination center



Interaction with CO<sub>2</sub>

### Interaction with H<sub>2</sub>O

- Oxygen vacancies (V<sub>O</sub>) offer *chemical adsorption sites for CO<sub>2</sub>*
- V<sub>o</sub> favours the dissociation of CO<sub>2</sub> on the surface of TiO<sub>2</sub>
- Oxygen deficient surfaces are more thermodynamically favourable to adsorb CO<sub>2</sub> than defect-free surfaces
- Defect sites promote the adsorption and dissociation of water on the surface of TiO<sub>2</sub>
- Under illumination and in the presence of CO<sub>2</sub> and H<sub>2</sub>O vapour, more bridged HCO<sub>3</sub> and HCOOH species were formed on defective TiO<sub>2</sub> than on defect-free TiO<sub>2</sub>, according to an in situ DRIFTS study

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Optimization of defects, band gap, band position

*CH*<sub>4</sub> *yield* of 80.35  $\mu$ molg<sup>-1</sup>h<sup>-1</sup> / 30 h stability

#### Materials Today 20 (2017) 507-515

# **High-Rate Solar-light Photoconversion of CO<sub>2</sub> to Fuel** : Controllable Transformation from C<sub>1</sub> to C<sub>2</sub> Products

Energy Environ. Sci., 11 (2018) 3183 - 319















R.E. Schaak et al., "A total synthesis framework for the construction of high-order colloidal hybrid nanoparticles," *Nature Chemistry* **2012**, *4*, 37-44.

> CO<sub>2</sub>, Water, and Sunlight to Hydrocarbon Fuels: A Sustained Sunlight to Fuel (Joule-to-Joule) Photoconversion Efficiency of 1% Energy & Environmental Science (2019) 12, 2685 - 2696

Energy &

Science

Environmental



**Figure 1:** HR-TEM image of  $Cu_{1.00\%}$ -Pt-<sub>0.35\%</sub>-BT (inset showing SAED pattern of region encircled with yellow)

- The presence of *Pt and Cu-Pt alloys* is confirmed from the d-spacing of the *lattice fringes*, which are 0.23 nm (111) for Pt and 0.22 nm and 0.18 nm (111) for Cu-Pt
- \* XPS analysis also confirms the presence of Cu-Pt nanoparticles



Fig. 2 Comparative methane evolution, UV-visible DRS, and PL spectra. (a) Effect of sample composition and structure on CO2 photoconversion to CH4 yield. (b) UV-visible diffuse reflectance spectra (DRS) indicate an enhancement in optical absorption with Cu–Pt nanoparticles, with absorption increasing with Cu content, and (c) PL spectroscopy indicates a better charge separation is observed for bimetallic deposited blue titania sample.

# Cu<sub>2</sub>O-reduced-titania (RT) Z-scheme Heterostructure

**Synthesis Scheme** 



## **☐** Why Cu<sub>2</sub>O-RT heterostructure is unique????

- Interface is formed at electrons enriched defected core (white arrows fig(a)) as shown in the HR-TEM fig(a-c)
- Unlike core-shell structure, both catalyst can take part in the reaction
- Enriched electron density at interface avoids the Cu<sub>2</sub>O oxidation

#### **HR-TEM of RT-Cu<sub>2</sub>O**



### Applied Catalysis B: Environmental, 279 (2020) 119344

# Photocatalytic CO<sub>2</sub> Reduction over Cu<sub>2</sub>O-RT Z-scheme



Fig. (a) Comparison of  $CH_4$  evolution over a 6 h test for P25, RT, and RT- $Cu_x$ ; (b)  $CH_4$  evolution for variable reduction degree from slightly reduced RTs- $Cu_{0.75}$  to highly reduced RTh- $Cu_{0.75}$ .

- Better charge separation, increased light harvesting and strong redox potential translated to 0.13% photoreduction of CO<sub>2</sub>, fig(a)
- Optimum metal deposition (Cu = 0.75%) and defects ( $Ti^{3+}, V_0$ ) are essential for better performance, fig(b)

Synergistic effects of Cu<sub>2</sub>O Z-scheme with RT, guarantee the efficient CO<sub>2</sub> photoreduction

# **Stability for Photocatalytic Reduction of CO<sub>2</sub>**



- Proposed stability is evident in CO<sub>2</sub> photo-reduction fig(a) for 7 cycles (total 42 h)
- Oxidizing h<sup>+</sup> on Cu<sub>2</sub>O are quenched by e<sup>-</sup> from the RT as shown in fig (b)
- Amorphous interfacial layer ensures the supply of e<sup>-</sup> to quench Cu<sub>2</sub>O oxidizing h<sup>+</sup>

Charge transfer/separation and CO<sub>2</sub> reduction performance/stability confirms

stable Z-scheme heterostructure of RT and Cu<sub>2</sub>O

# Topic 2: Hydrogen Production by Hybridization of Microbial Electrolysis Cell and photoanode

STEAM METHANE REFORMING



SMR







• Microbial Fuel cells (MFCs)



• MFCs produce electricity with decomposing organic matter (in wastewater).

## Exoelectrogens





PECs



Fig 7. Power overshoot on MFCs

Table 1. Summary of causes of power overshoots in previous studies.		
Causes of power overshoot	Previous study	Ref.
Increase of internal resistance,	leropoulos	[9]
ionic depletion	et al., 2010	
Increase of internal resistance	Liu et al., 2011	[10]
Inadequate scan time of discharge test, insufficient	Winfield et al.,	[11]
inoculation of bioanode, limited organic concen- tration	2011	
Inadequate scan time of discharge test	Watson and Logan, 2011	[12]
Poor acclimation of microbes	Hong et al., 2011	[13]
Poor acclimation of microbes	Zhu et al., 2013	[14]
Lack of anodic capacitance	Peng et al., 2013	[15]
High substrate utilization resistance,	Nien et al., 2011	[16]
high electron transfer resistance		
Electron Depletion	This study	
(deduced as regards of causes from previous stud-		
ies, because all causes affect to production of elec- tron)		

- The power overshoot is defined as a "doubling back" of the voltage and power curves toward lower currents, rather than increasing to the expected higher current.
- Ax-Cy: A means anodes and x is the number of anodes C means cathodes and y is the number of cathodes (A3-C2, A4-C2, and A4-C3 MFC).
- Power overshoot is eliminated by using more anodes.





- MECs under dark showed a gas production rate 1096.809 mmol m<sup>-3</sup> h<sup>-1</sup> H<sub>2</sub> and 97.816 mmol m<sup>-3</sup> h<sup>-1</sup> CO<sub>2</sub>.
- hybrid MECs showed a gas production rate 1434.268 mmol m<sup>-3</sup> h<sup>-1</sup> H<sub>2</sub> and 84.894 mmol m<sup>-3</sup> h<sup>-1</sup> CO<sub>2</sub>. (30.76% H<sub>2</sub> production rate improvement)
- Columbic efficiency of MECs under dark showed 26.99% and that of hybrid MECs showed 35.30%. (8.31% improvement)

Su-II In and et al. *Energies* 11, 3184 (2018)

# **Topic 3: C-14 Powered Dye-Sensitized Betavoltaic Cells**



Chemical Communications (2020) 56, 7080-7083





#### Necessity of betavoltaic battery

- With the recent development of portable devices and electric cars, the demand of long-lasting batteries increases.
- Present lithium-ion batteries have constraints of frequent replacement, periodic charging and energy efficiency.
- The big problem is the increasing amount of battery waste as the days go by.
- The study of the next generation of long-lived batteries is very important.
- Betavoltaic batteries are attracting a lot of attention as one of the next-generation batteries. Beta electrons from implanted radioactive isotopes collide with semiconductors and generate electricity.
- It can supply the power by its own for a long time.
- (Low efficiency, complicated fabrication process and high cost are disadvantages)



#### **Technology trend**

Betavoltaic battery technology: generation and growth



Structure and principle of betavoltaic cells using radioisotope quantum dot nanoparticles



- Radioisotope carbon quantum dot electrode
- C-14 nanoparticles in the cathode have dual role as counter electrode material and radioactive source which simplify the device architecture.
- In addition, it improves performance by increasing the degree of radiation integration through C-14 nanoparticle as a beta source.

Radiation absorber (dye)

- There are **enhanced electron generation and transfer** by utilizing the N719 dye in the radiation absorber.
- The simple fabrication process **brings 20~30% cost savings**.

[Fig. 5] Schematic diagram of DSBC using radioisotope quantum dot nanoparticles.

Voltage, current and efficiency analysis of betavoltaic cells using radioisotope quantum dot nanoparticles



- The DSBC showed an efficiency of 0.48% with the short-circuit current density (J<sub>sc</sub>) 12.75 nA/cm<sup>2</sup>, and the open-circuit voltage (V<sub>oc</sub>) 29.2 mV.
- The DSBC generated 3.2 × 10<sup>4</sup> times more mobile electrons than those generated only by beta radiation due to the electron-impact multiple ionization or secondary ionization.
- The DSBC system using the N719 dye had a maximum power density of 0.095 nW/cm<sup>2</sup> with an energy efficiency of 0.48% because of the metal-ligand charge transition(MLCT) and difference of degrees of protonation compared to organic dye(SQ2) and N3.

#### Feature point of this technology



Stability test of betavoltaic cells using radioisotope quantum dot nanoparticles

[Fig. 10] Stability of the DSBC-N719 device: (a) J<sub>sc</sub> and V<sub>oc</sub> and (b) efficiency.

The stability of a DSBC-N719 over 10 hours period was measured to be 12.75 nA/cm<sup>2</sup> at 29.2 mV.

The fluctuation in  $J_{sc}$  and  $V_{oc}$  may be due to a variable beta electron emission rate.

The internal structure and design of the cells are suitable and stable.

#### 2020년 7월 16일 목요일 021면 전국

# 충전 없이 평생 쓰는 배터리 나온다

동위원소 '탄소-14'로 구조 단순화

반감기 5730년…반영구사용가능

우주·의료 분야 차세대 전원 기대

염료로 대체했다. 또 베타선을 방출하는 동위 원소 '탄소-14'를 적용해 기존 베타전지가 가진 복잡한 구조를 단순화했다. 탄소-14를 나노입 자로 만들어 에너지 밀도를 높였다. 염료감응 베 타전지 성능실험을 통해 탄소-14에서 방출된 전자 대비 3만2000배의 전자를 생성하며 10시 간 동안 안정적으로 전력을 생산한다는 것을 관 찰했다. 특히 베타전지에 사용된 탄소-14는 약 5730년의 반감기를 가지고 있어 상용화에 성공 하면 반영구적 수명을 가질 것으로 예상된다.

이번 연구는 베타전지분야에서 새로운 구조 와 방사선 흡수체에 적용할 수 있는 물질의 다 양성을 제시했다는데 의미가 있다. 연구팀은 후속연구로 베타전지 효율을 실용화 수준까지 끌어올릴 계획이다. 실용화에 성공하면 우주와 심해 등 극한 환경뿐 아니라 의료분야 차세대 전원으로도 활용 가능하다.

인수일 교수는 "값싼 염료를 적용, 새로운 베 타전지 개발에 성공했다는데 의미가 있다"면 서 "아직 풀어야 할 숙제가 많지만 안전하고 저 렴한 염료감응 베타전지 개발에 노력하겠다" 고 밝혔다. 인 교수는 2010년부터 2년 동안 미 국 펜실베니아주립대 화학과, 재료공학과 박사 후 연구원으로 활동한 뒤 DGIST 에너지공학전 공 교수로 부임해 차세대 에너지 분야를 집중 연구하고 있다.

DGIST, 염료감응 베타전지 개발



⁵ 전자신문

K3300 ELX Solar Cell Micro Imaging Sys

충전 없는 배터리 시대

앞당긴다.세계<u>최초</u>

'염료감응 **비**타전지' 개발

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대학 연구팀이 별도 충전 없이 반영구적으로 사용 가능한 염료감응 베타전지를 개발했다. 극한 환경이나 의료 분야 차세대 전원으로 활 용될 수 있을 것으로 기대된다.

대구경북과학기술원(DGIST·총장 국양)은 인수일 에너지공학전공 교수팀이 베타선에 염 료가 반응하는 원리를 응용해 값싸고 안전한 반영구베타전지를 최초로 개발했다고 15일 밝 혔다.

기존 전지는 우주, 심해 등 극한 환경에서 사 용하기에 전력과 장기 안정성이 떨어지고, 수 명도 짧다. 게다가 잦은 교체 주기에 따른 폐기 물 발생으로 환경 문제를 일으킨다.

베타전지는 방사성 동위원소를 원료로 이용 하는 차세대 전지 중 하나다. 방사성 동위원소 에서 방출된 베타전자가 방사선흡수체인 반도 체에 충돌하면서 전기를 생산하는 원리다.

베타선은 인체 유해성과 투과도가 낮아 높은 안전성을 띤다. 외부 동력원 없이 자체 전력 생 산이 가능해 별도의 충전이 필요 없다. 수명은 방사성 동위원소의 반감기와 비례하기 때문에 교체 주기가 길다. 미국과 러시아 등 세계 주요



인수일 DGIST 교수팀이 개발한 염료감응 베타전지.

국가가 베타전자 연구에 각축을 벌이고 있는 이유다. 하지만 소재가 비싸고 복잡한 제작 공 정 때문에 대량생산이 쉽지 않다.

인 교수팀은 기존 베타전지에서 방사선흡수 체로 사용된 값비싼 반도체 물질을 '루테늄(Ru thenium: 주기율표의 중앙에 위치하는 전이금 속으로, 백금족 금속의 하나)' 계열의 'N719'

**CHEMISTRY**NORLD

#### Solar cell concept used to enhance nuclear battery

DGNNDG

대구경북과학기술원

ses a ruthenium dve to absorb beta radiation



## **Topic 4: Nano-Bio Hybrid Science and Technology**



## Drug misuse/overdose and environmental pollution







# Introduction

## Acupuncture therapy



- The invention of acupuncture as a therapeutic treatment is traced as far back as 6000 B.C., originating with the insertion of sharpened stones at specific acupuncture points.
- An acupuncture device is made by metal acupuncture needles, including those of gold, silver, copper, and stainless steel.
- We hypothesize that an increase in needle surface area with no significant variation in needle diameter may lead to increased interactions of surrounding tissue, leading to enhanced acupuncture stimuli. -> autoimmune system

## **Brief History of Acupuncture**



## Hierarchical Micro/Nano-Porous Acupuncture Needles Offering Enhanced Therapeutic Properties

Scientific Reports 2016, 6, 34061



Enhanced Therapeutic Treatment of Colorectal CancerUsingSurface-ModifiedNanoporousAcupunctureNeedlesScientific Reports 7 (2017) 12900

## - SEM (Scanning Electron Microscope)



- (a) Conventional stainless steel needle
- (b) Porous anodized needle
- (c) Enlarged image of (b)
- (d) Cross-sectional image of the porous anodized needle.

Insets of (a, b) show needle tips.

- One can see that the pores are conical in shape, tapering in size from 3.0 μ m at the surface to 0.05 μ m, with a cone depth varying from 1.0 to 2.6 μ m.
- No formation or loss of any element in the acupuncture needle is detected, assuring that electrochemical anodization doesn't change the chemistry of the acupuncture needle.

## II. Neuronal activity



Conventional and PAN groups show increased neuronal activity to needle stimulation (after) compared to before stimulation (\*p < 0.05). In addition, PAN group shows a significant increase in stimulationinduced neuronal activity compared to conventional group (#p < 0.05).



- Early stage colorectal cancer (CRC) gene expression analysis and aberrant crypt foci (ACF) formation



- ACF increase in crypt multiplicity with time, and thus are accepted as a predictor of tumor progression
- The mean number of total ACF in I2 was  $31.0 \pm 12.5$
- The ACF count dramatically decreased to 0.8 ± 1.5 ACF/colon in I5, demonstrating that PN at HT7 acupuncture significantly inhibits AOMinduced colonic AC formation
- The reduced ACF number is, presumably, closely related to the recovered expression of genes altered by AOM exposure.

Average number of ACF found in each group. The experimental conditions of initiation groups I2-I6, all at early stage CRC, are as follow: I2 = positive control(AOM injection only); , I3 = CN acupuncture treatment at HT7 point; I4 = CN acupuncture treatment at SI5 point; I5 = PN acupuncture treatment at HT7 point; and ; I6 = PN acupuncture treatment at SI5 point. - Analysis of tumor size and cancer biomarker (CEA)



(a) Average size of tumors identified in the colon of animals from each experimental group.

(b) Expression of the circulating colorectal cancer biomarker, CEA. All animals were treated by acupuncture at acupoint HT7 every day for 40 weeks after the last azoxymethane (AOM) injection. M8, positive control; M9, treated with the conventional needle (CN); M10, treated with the nano-porous needle (PN). Maturation groups M7-M10, all at late stage CRC, have experimental conditions: M7 = negative control (no treatment); M8 = Positive control (AOM injection only); M9 = CN acupuncture treatment at HT7 point; and M10 = PN acupuncture treatment at HT7 point.

- The average tumor size observed in M10 was significantly decreased from that of M8
- Figure 3b shows a significant decrease in the average CEA concentration in M9 and M10 compared to M8, providing further evidence as to the efficacy of acupuncture treatment

[첨부1\_그림2] 국문(좌) 영문(우) 박스에 포장된 시제품 SNP침 모습



[첨부1\_그림1] 국문(좌) 영문(우) 쌈지에 포장된 시제품 SNP침 모습





[제품개발] Smart Nano Porous Needles (SNP Needles, SNP침) 시제품 이미지

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# Thank you!

Q&A